## Recent developments in sulphur bentonite fertilisers

C.C. Boswell<sup>1</sup>, W.R. Owers<sup>2</sup>, B. Swanney<sup>1</sup>

<sup>1</sup> Invermay Agricultural Research Centre, Private Bag, Mosgiel, New Zealand <sup>2</sup> Industrial Processing Division, D.S.I.R., Private Bag, Petone, New Zealand

The effectiveness of elemental sulphur (S?) as a fertiliser is dependent on the rapidity with which it is oxidised by micro- organisms to plant available  $SO_4$ -S form. Early field studies showed that particles > 0.25 mm were ineffective for sub- clover pastures (1). However, finely divided S? particles form an explosive mixture with air. Experimental S-bentonite prills made by mixing molten sulphur with sodium bentonite represents an attempt to produce safe and effective high analysis S fertiliser.

## Methods

Laboratory measurements were made of both the rate of dispersion of S?-containing fertilisers in water and the distribution of S? particle sizes. The rate of release of SO<sub>4</sub>-S from a range of S? fertilisers, applied at 50 kg S/ha, was subsequently measured from plant SO<sub>4</sub>-S levels in a ryegrass-white clover pasture on a Warepa soil (which has low anion retention properties) at Invermay Agricultural Research Centre, N.Z. (mean annual rainfall 687 mm), during the 1982 and 1983 growing seasons.

## **Results and Discussion**

Experimental S-bentonite materials (made in N.Z. by the Industrial Processing Division, D.S.I.R.) have shown that increasing the bentonite content from 5 to 40% progressively increased the rate of dispersion in water and the fineness of the dispersed sulphur (Table 1). There was some release of SO-S from the 40% bentonite material within 29 days after the start (DAS) of a field experiment. By 145 DAS there was a linear relationship between bentonite content and plant SO<sub>4</sub>-S levels. Control (nil S) levels were 0.55% and 117% respectively at 29 and 145 DAS while those from gypsum fell from .134% to 100%.

## Table 1 Effect of bentonite content on prill disintegration, S? particle size and sulphate release

Disintegration time 1 disintegration 101 801				Cumulative % of after dispersal . < .15 pm	particles in water < .5 mm	Plant SO <sub>4</sub> -S 29 DAS	levels (%) 145 DAS
10	secs	3	min	100	100	.070	.175
20 t	secs min	35	min min	87	100	.063	.143
12 10	min days	>4	days N/A	46	96 20	-055	.117
	Disi 10 20 1 12 10	Disintegra 2 disint 102 10 secs 20 secs 1 min 12 min 10 days	Disintegration 1 disintegra 10% 10 secs 3 20 secs 7 1 min 35 12 min 34 10 days	Disintegration time 1 disintegration 10% 80% 10% secs 3 min 20% secs 7 min 1 min 35 min 12 min 34 days 10 days 8/A	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Disintegration time Cumulative % of particles Plant SO <sub>4</sub> -S   % disintegration after dispersal in water 29 DAS   10% 80% < .15 mm

A range of alternative S fertilisers were evaluated in a separate experiment at an adjacent site for over 1 year. An experimental S-bentonite (30% bentonite) was included. There was a marked residual effect of the S-bentonite (plant  $SO_4$ -S = In contrast there was little effect from either gypsum (.093%) or a coarser (0:5-1.0 mm) fraction of S<sup>?</sup> (0.80%  $SO_4$ -S).

S-bentonite prills with at least 10% of sodium bentonite are high analysis S fertilisers which are stable for safe storage and application; readily disperse in water; and the S? is rapidly oxidised to sulphate. Despite the rapid oxidation they have a longer residual effect than can be expected from sulphate fertilisers.

1. Weir, R.G., Barkus, B., Atkinson, W.T. 1963 Aust. J. Exp. Agric. Anim. Husb. 3: 314-18.